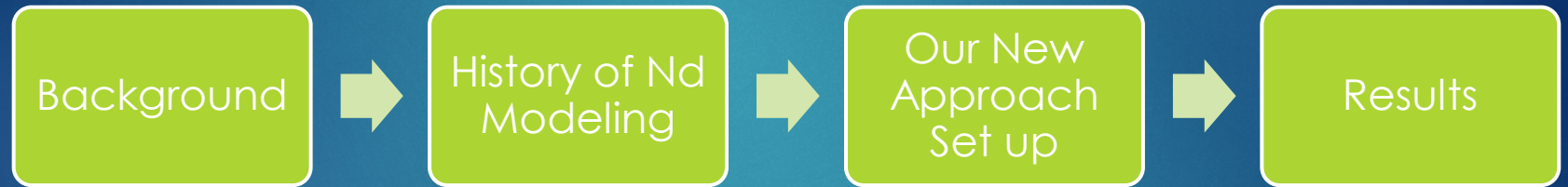


New Approach to Modeling Neodymium: Testing the Bottom-up hypothesis

MIA EL-KHAZEN

ANDREAS SCHMITTNER AND BRIAN HALEY



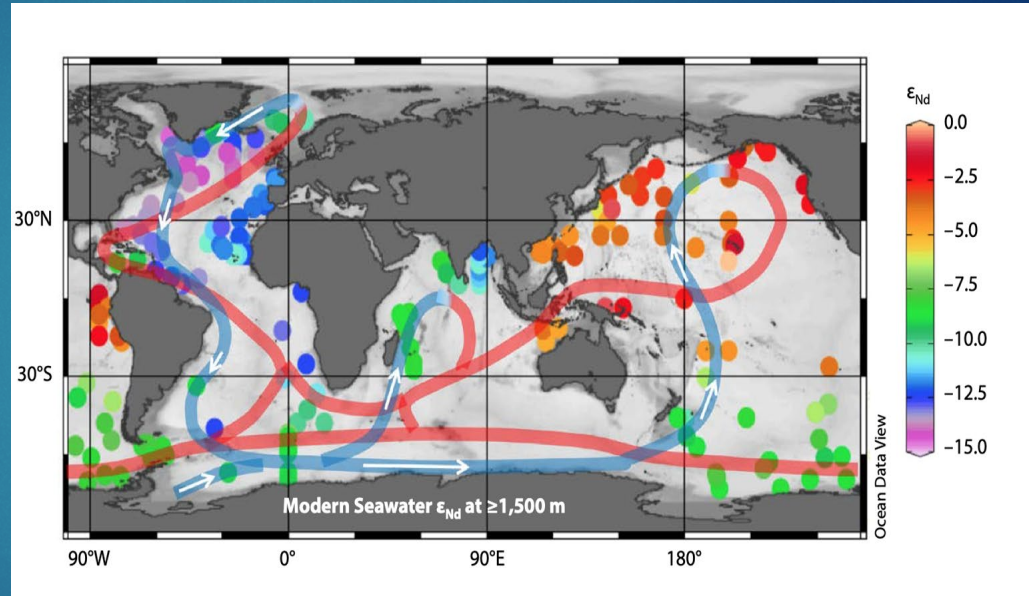
Background information

General information:

- ↓ Rare Earth Element
- ↓ Isotopes formed from alpha decay of Samarium (Sm)

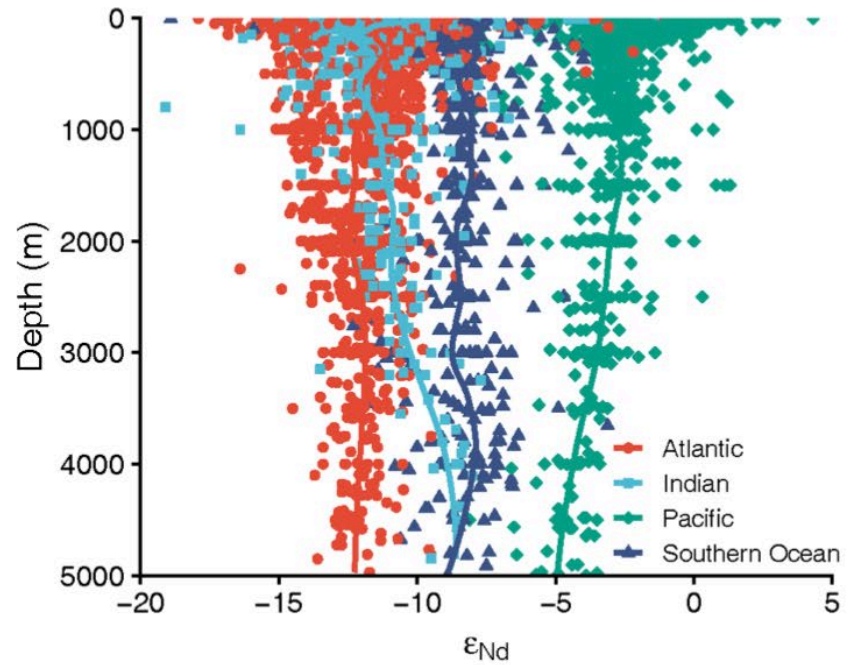
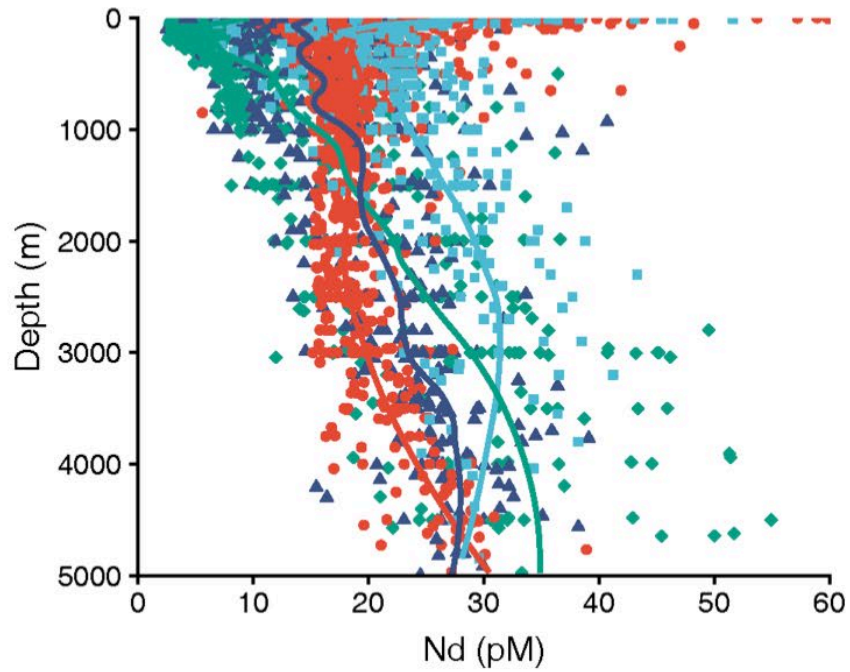
Why do we use ϵ_{Nd} as a tracer?

- ↓ Negligible biological fractionation
- ↓ Air-sea exchange
- ↓ Quasi-conservative

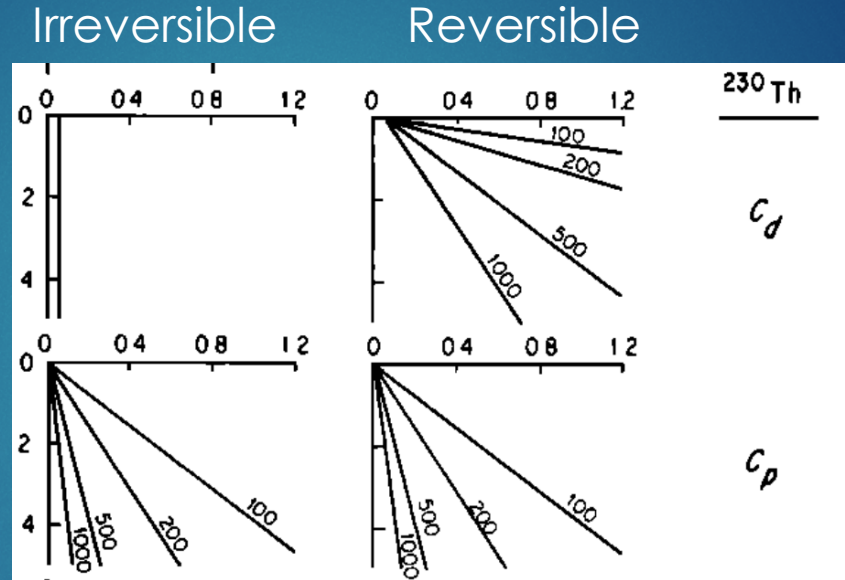
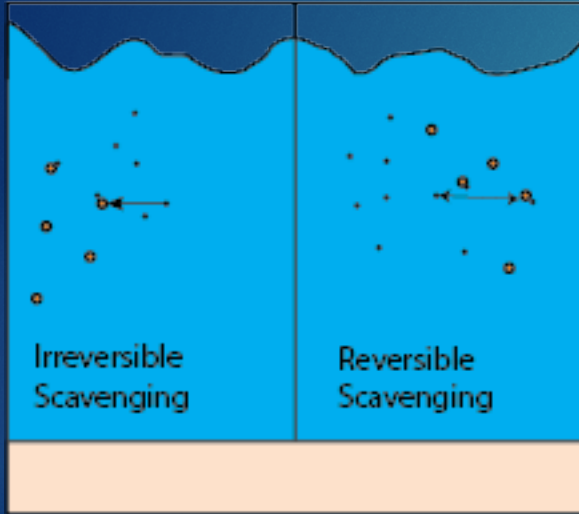


Tachikawa et al., 2020





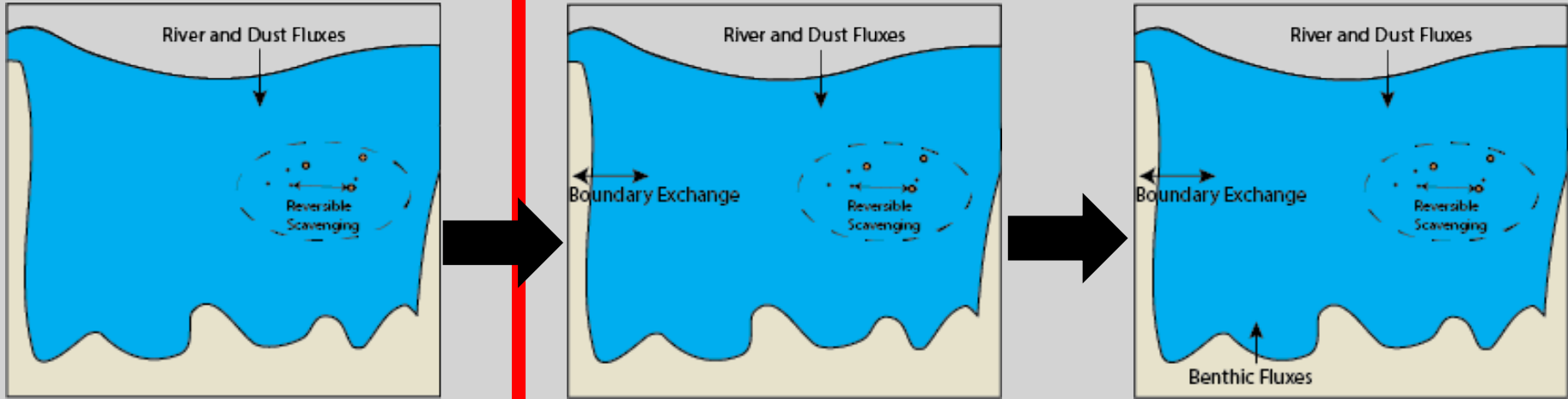
History of Neodymium modeling



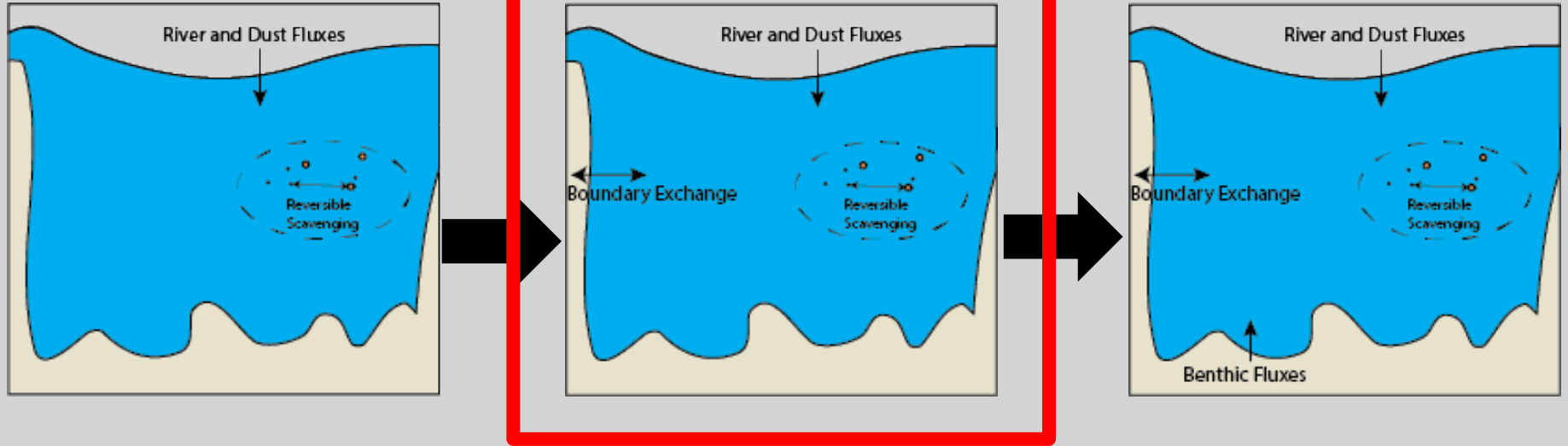
Bacon and Anderson, 1982

- Constant production in the water column
- No vertical diffusivity
- Constant sinking speed
- High C_p/C_d ratio

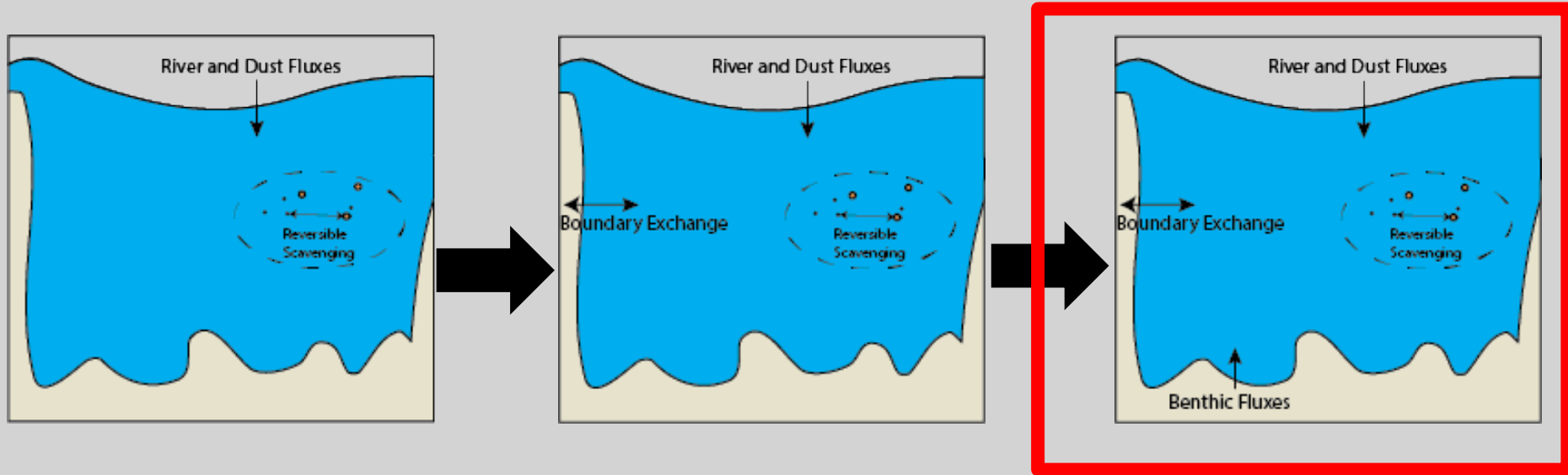
History of Neodymium modeling



History of Neodymium modeling

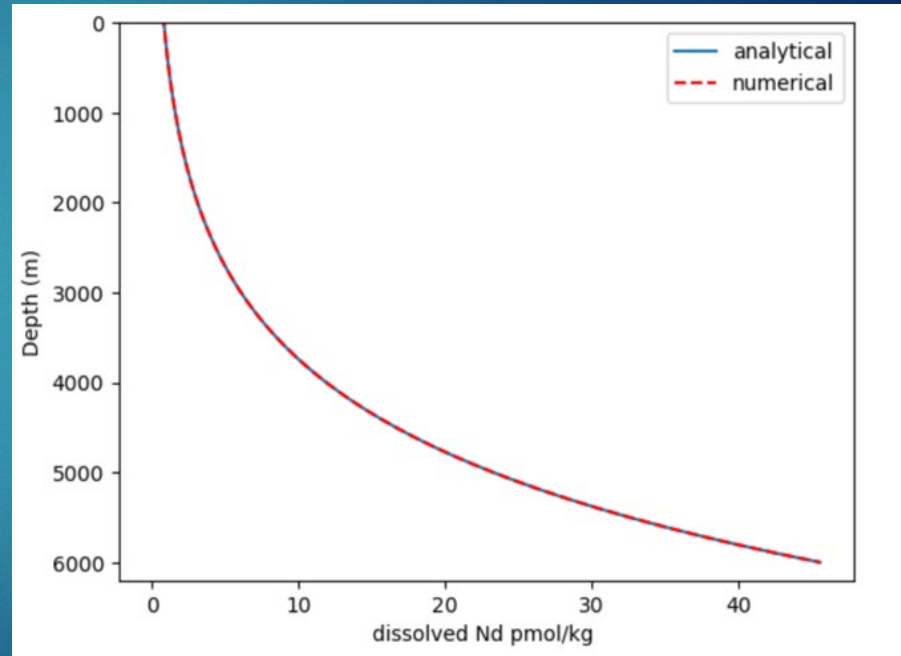
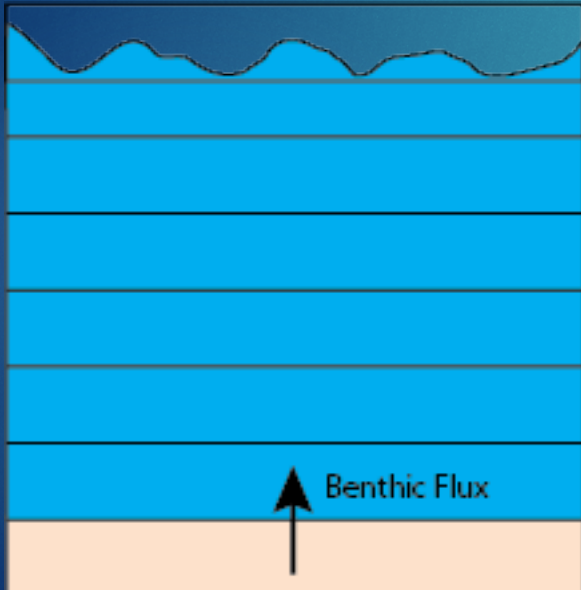


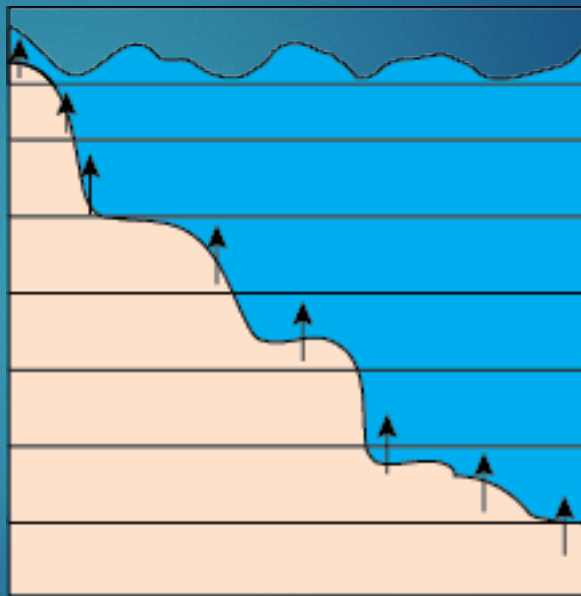
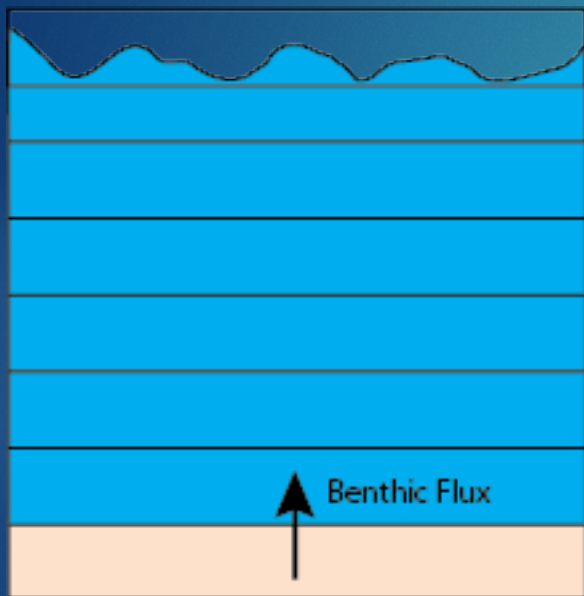
History of Neodymium modeling



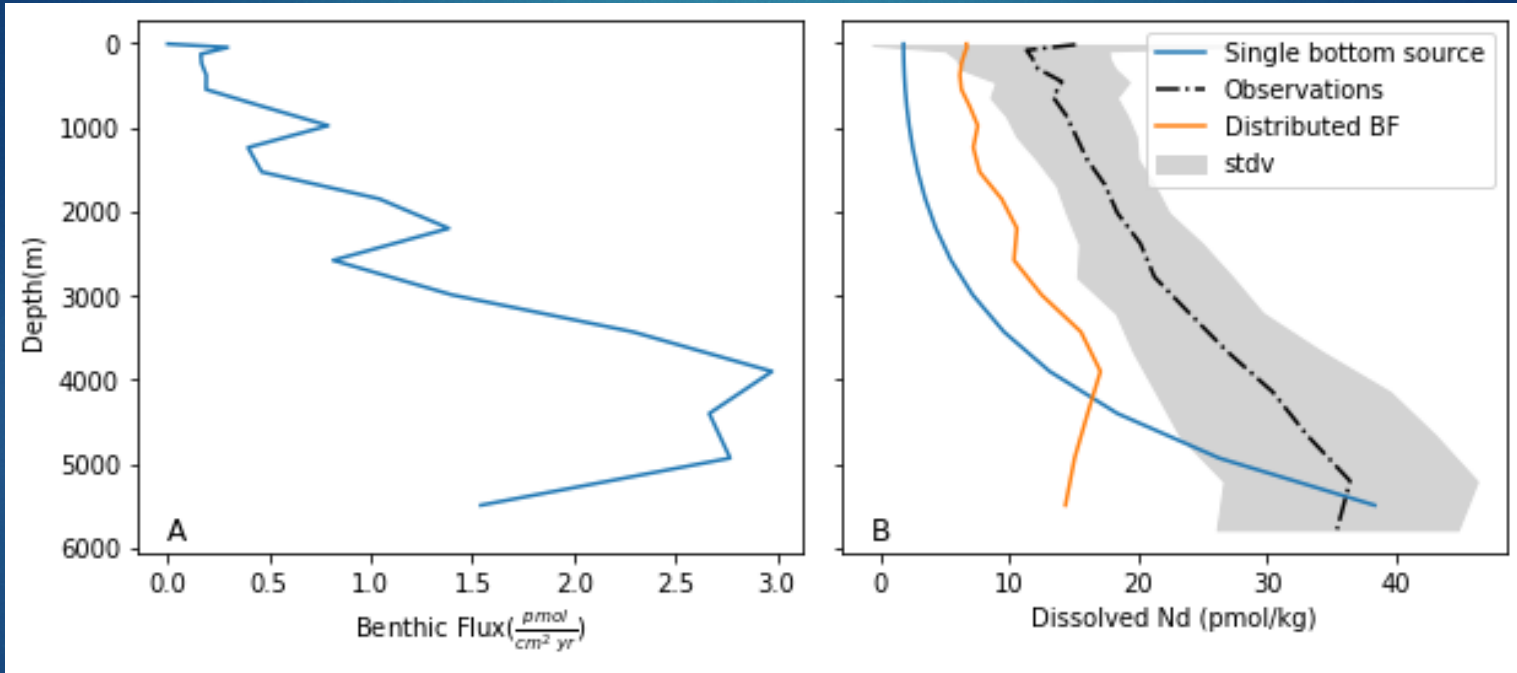
New Approach: Testing the bottom-up hypothesis

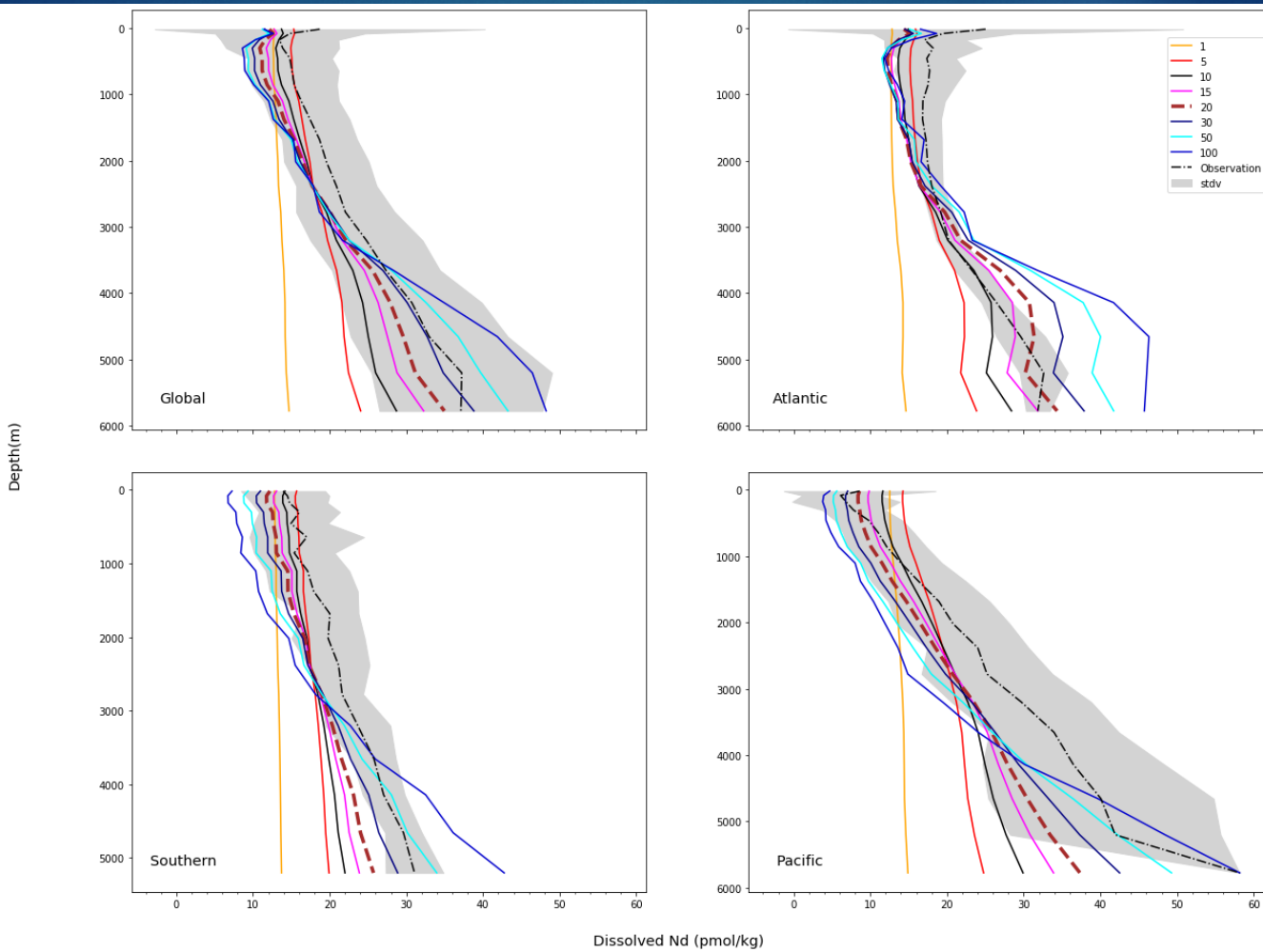
$$\frac{\partial c_d}{\partial t} = \underbrace{k_v \frac{\partial^2 c_d}{\partial z^2}}_{\text{Diffusion}} - \underbrace{k_p c_d}_{\text{Absorption}} + \underbrace{F_b}_{\text{Benthic Flux}}$$

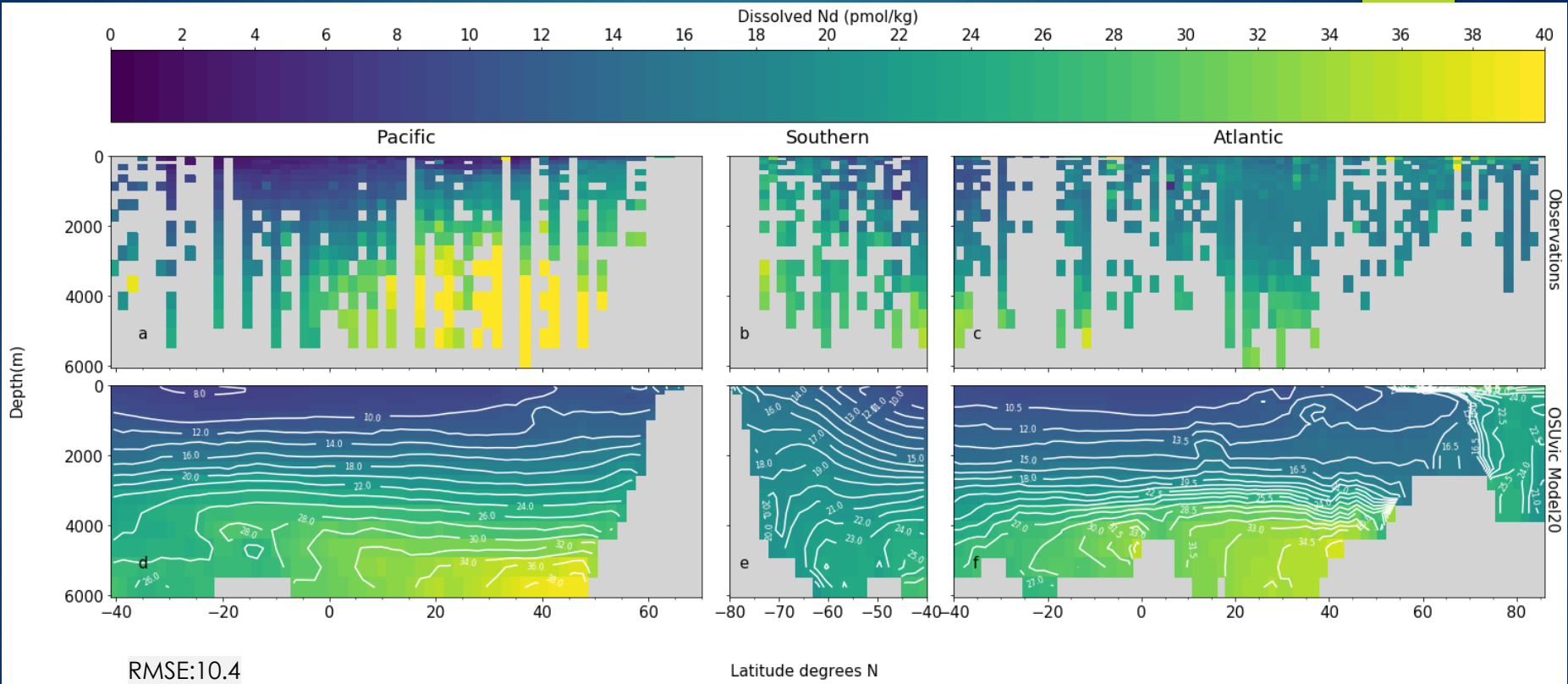




$$\frac{\partial C_d}{\partial t} = k_v \frac{\partial^2 C_d}{\partial z^2} - k_p C_d + F_b$$



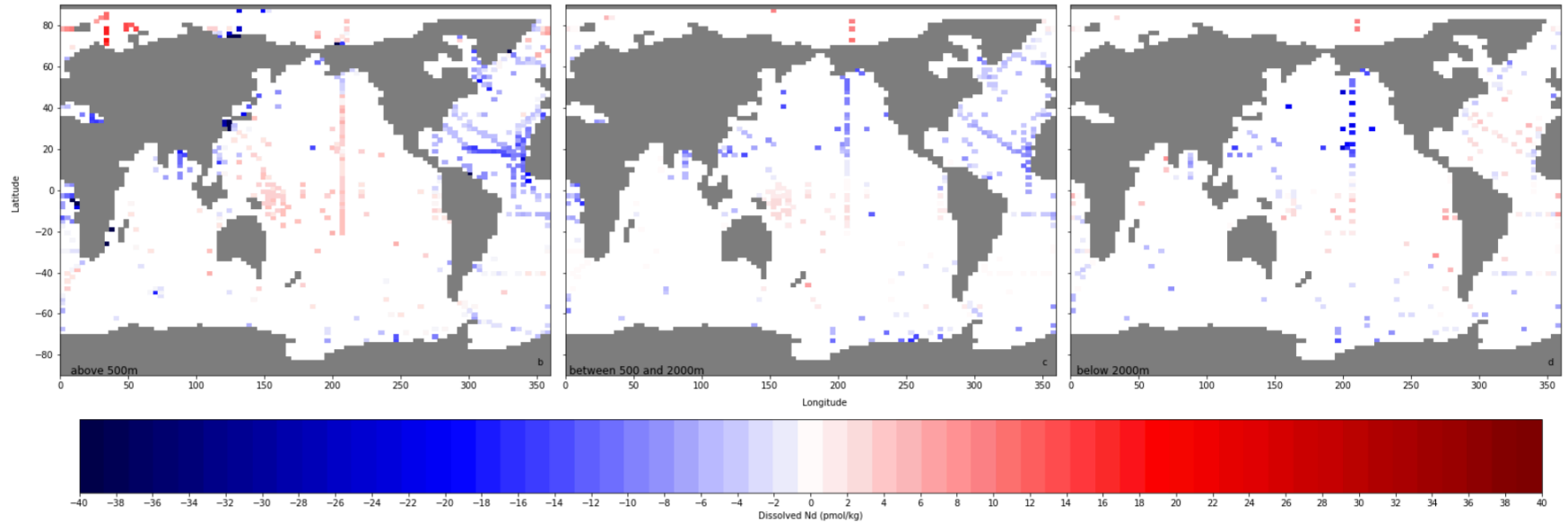




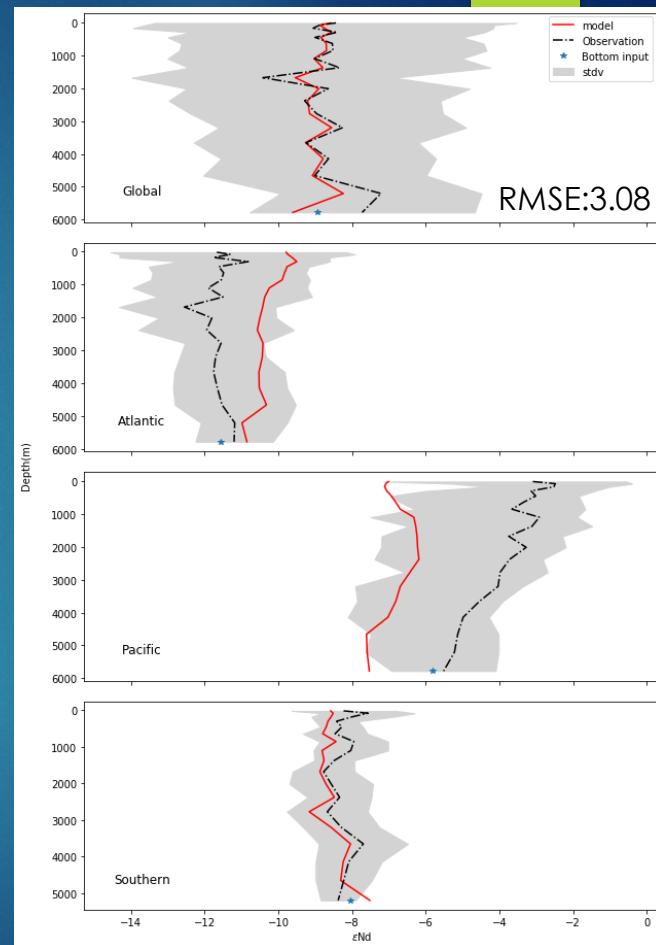
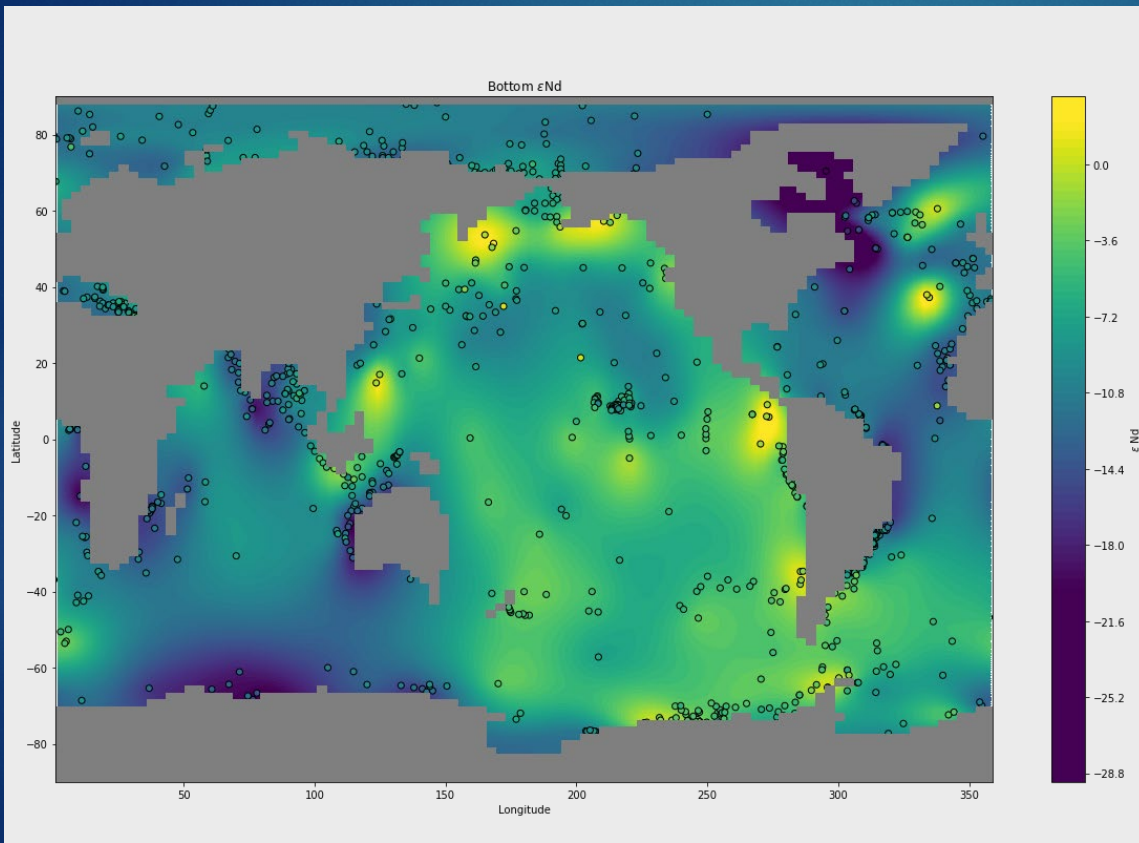
Above 500m

500m -2000m

Below 2000m



Future work



Some Key Points

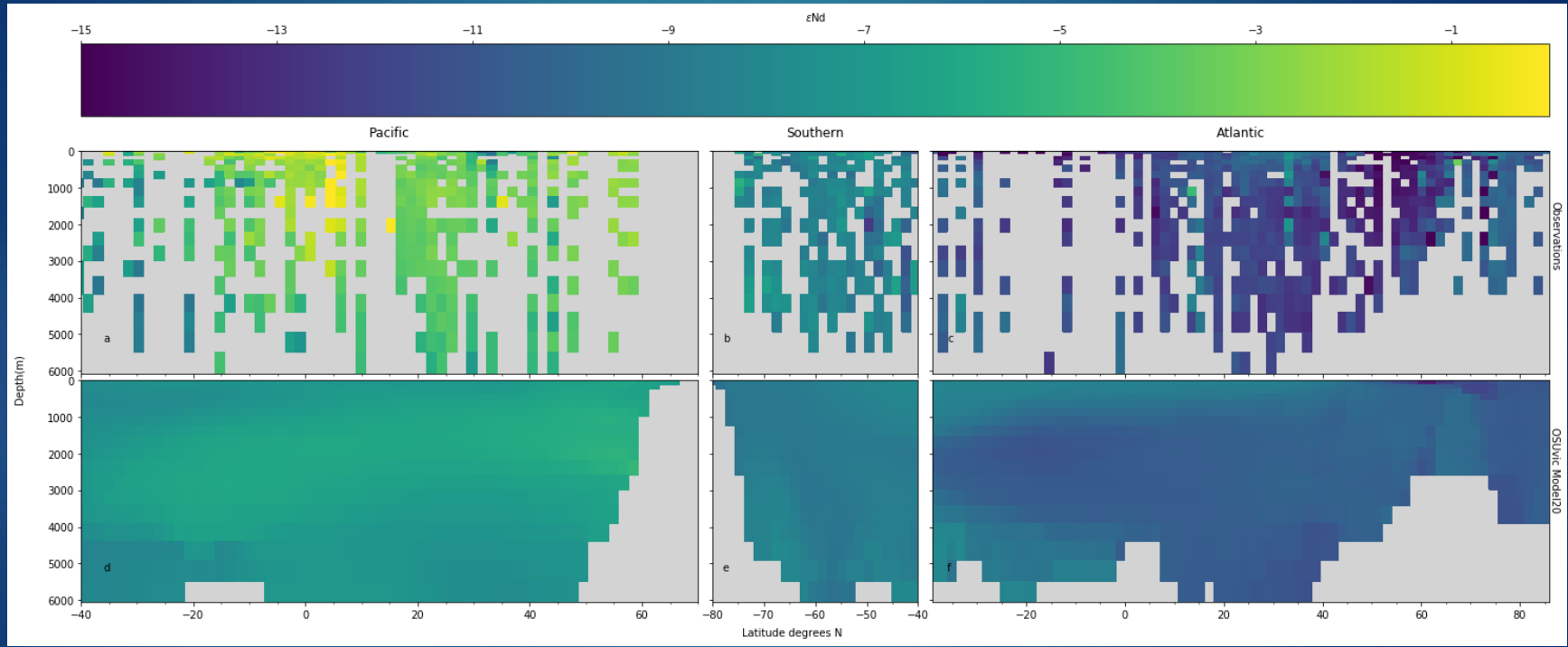
- A constant benthic flux, vertical diffusion, and irreversible scavenging result in dissolved Neodymium concentrations close to observations
- The bathymetry of the ocean, diffusion, and lateral transport lead to linearly increasing Neodymium concentrations with depth in the palagic ocean without the need for reversible scavenging
- Reversible scavenging does not play a major role in Neodymium distributions when using a bottom-up approach.

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University

Extra slides for Possible Questions

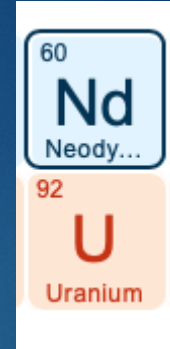


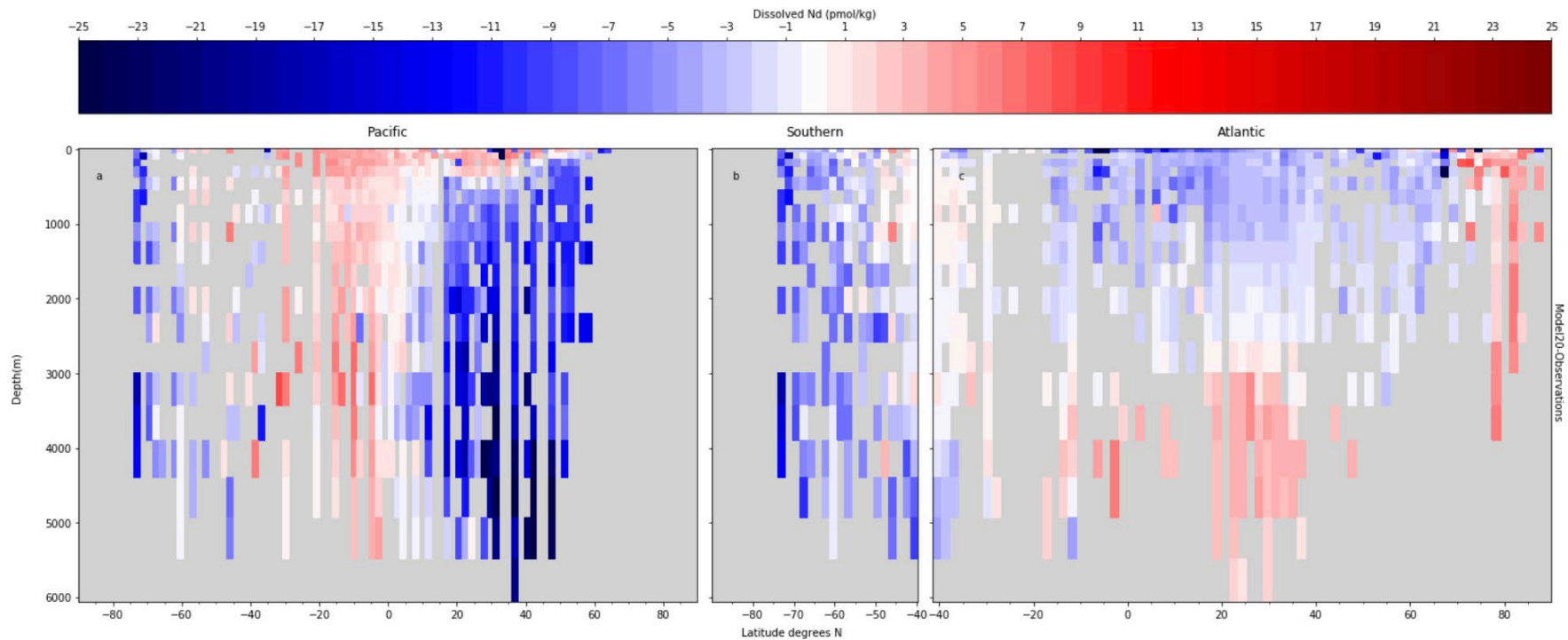
Background

- ↓ Rare Earth Element
- ↓ Isotopes formed from alpha decay of Samarium (Sm)
- ↓ CHUR: present-day Chondritic Uniform Reservoir 0.512638

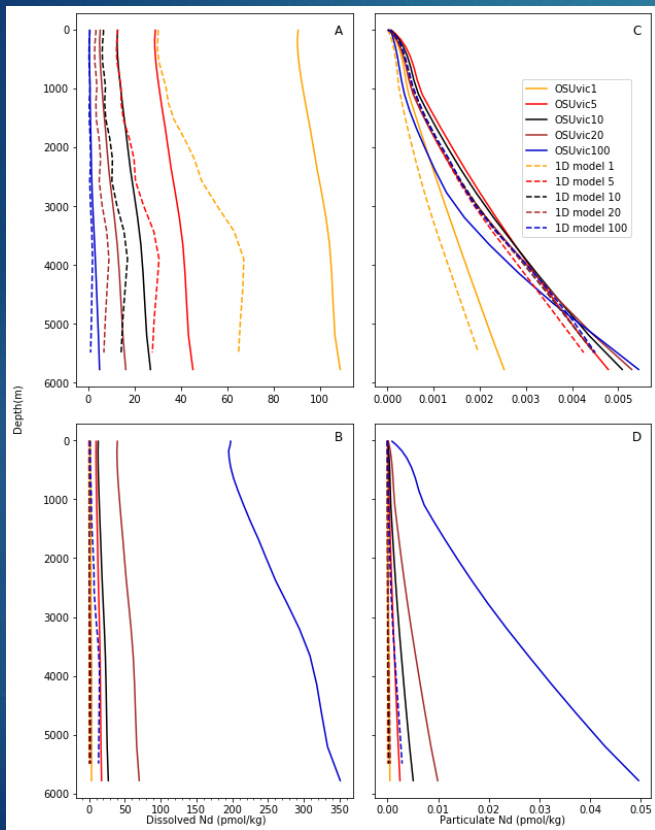
$$\epsilon_{\text{Nd}} = \left(\frac{(^{143}\text{Nd}/^{144}\text{Nd})_{\text{sample}}}{(^{143}\text{Nd}/^{144}\text{Nd})_{\text{CHUR}}} - 1 \right) \times 10^4$$

Jacobsen and Wasserburg, 1980

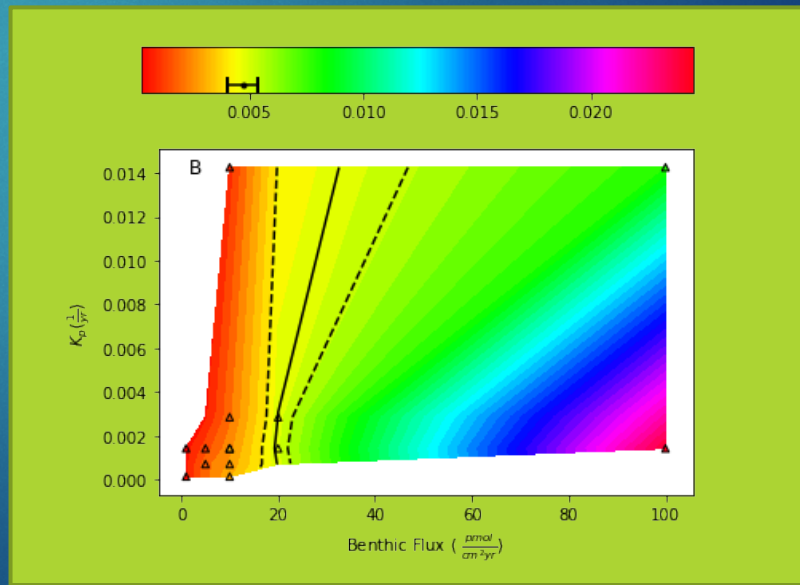


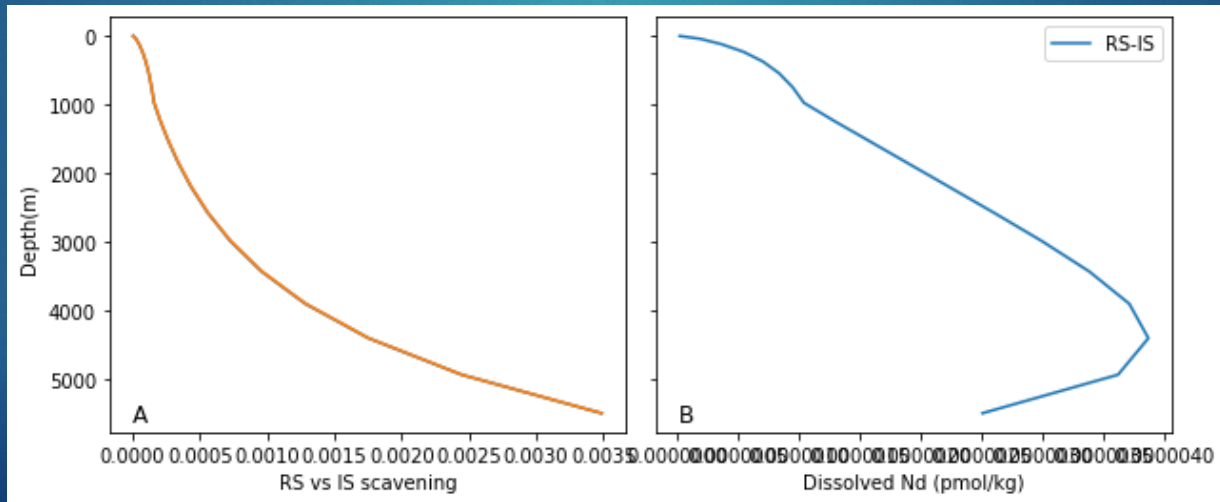
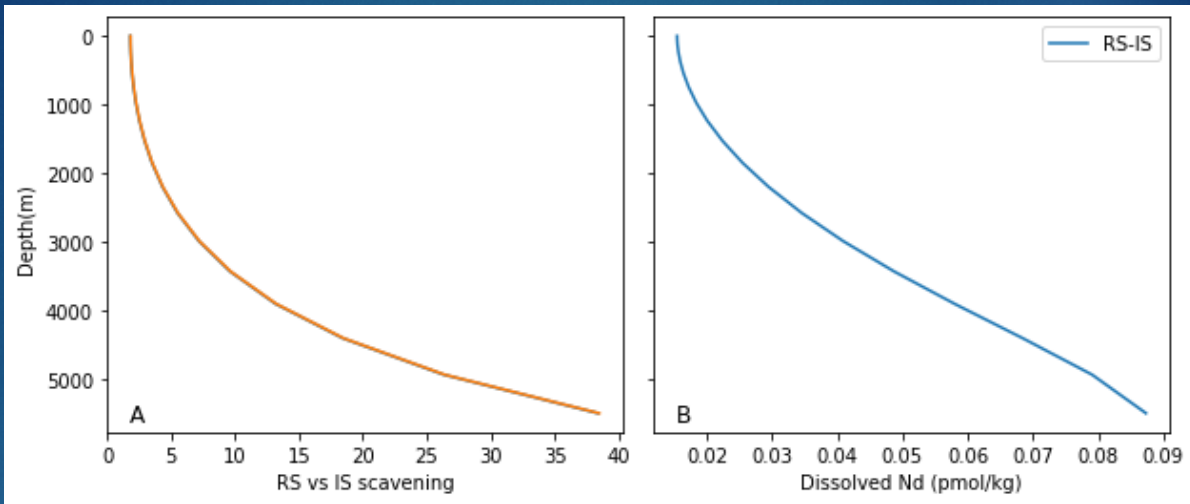


- ▶ Top: variable K_p
- ▶ Bottom: variable Benthic flux



BF (pmol cm ⁻² yr ⁻¹)	K_p (1/yr)
1	0.14×10^{-3}
5	0.71×10^{-3}
10	1.4×10^{-3}
20	2.9×10^{-3}
100	14×10^{-3}





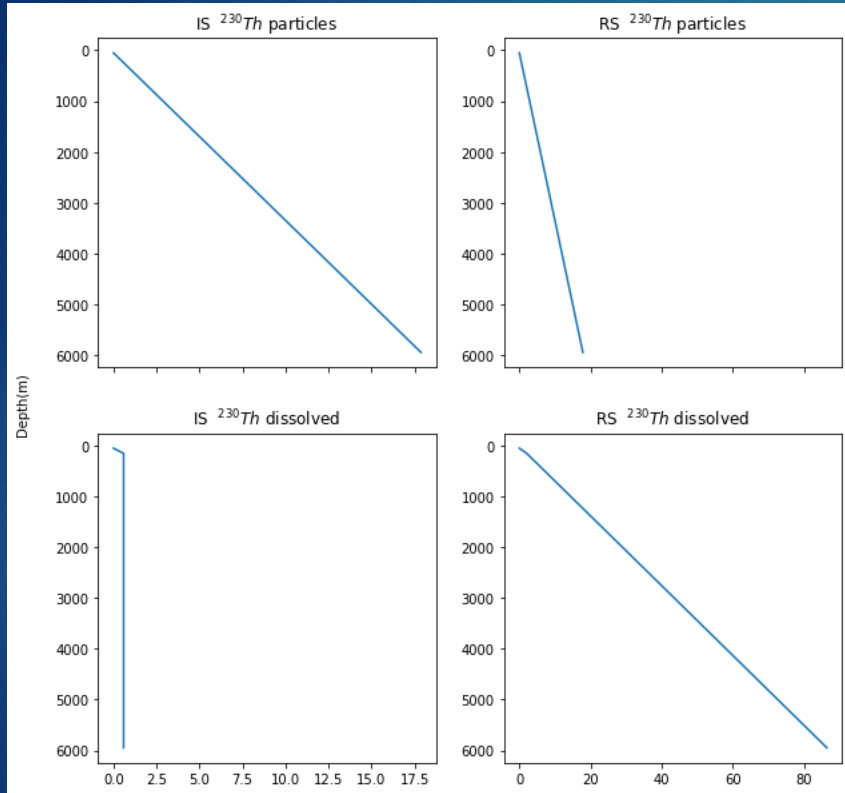


Figure S.1. The left side shows irreversible scavenging (IS) for both dissolved (bottom) and particulate (top) phase ^{230}Th , and the right shows the reversible scavenging (RS) results using the values shown in Table S.1.

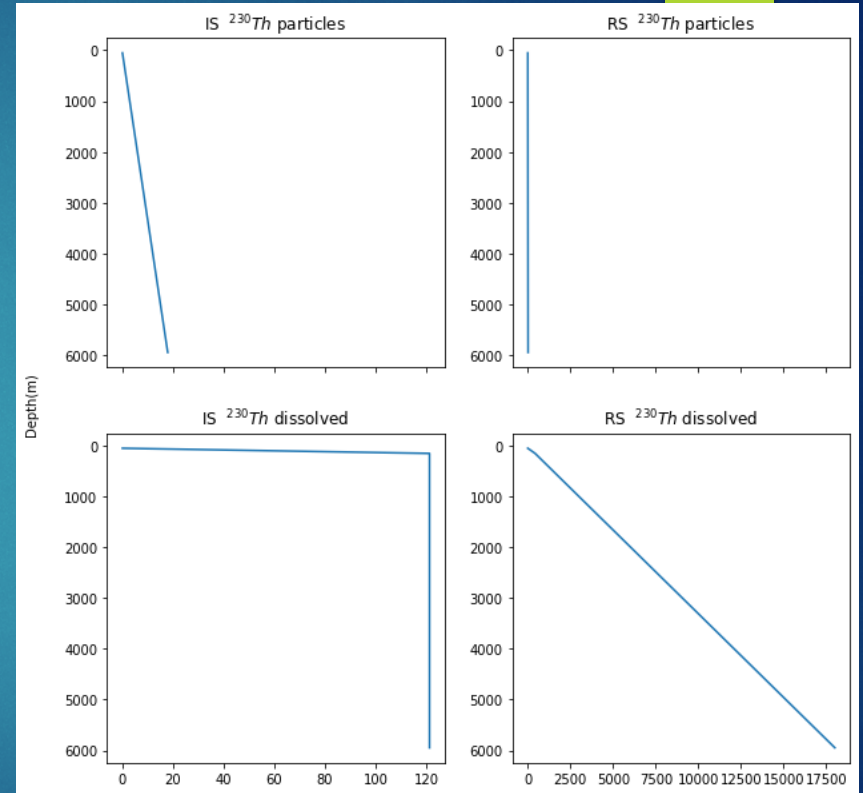


Figure S.2 Modified scavenging based on C_p/C_d ratio of 0.001. The left side shows IS results, and the right shows RS results. The particulate (top) and dissolved (bottom) concentrations.

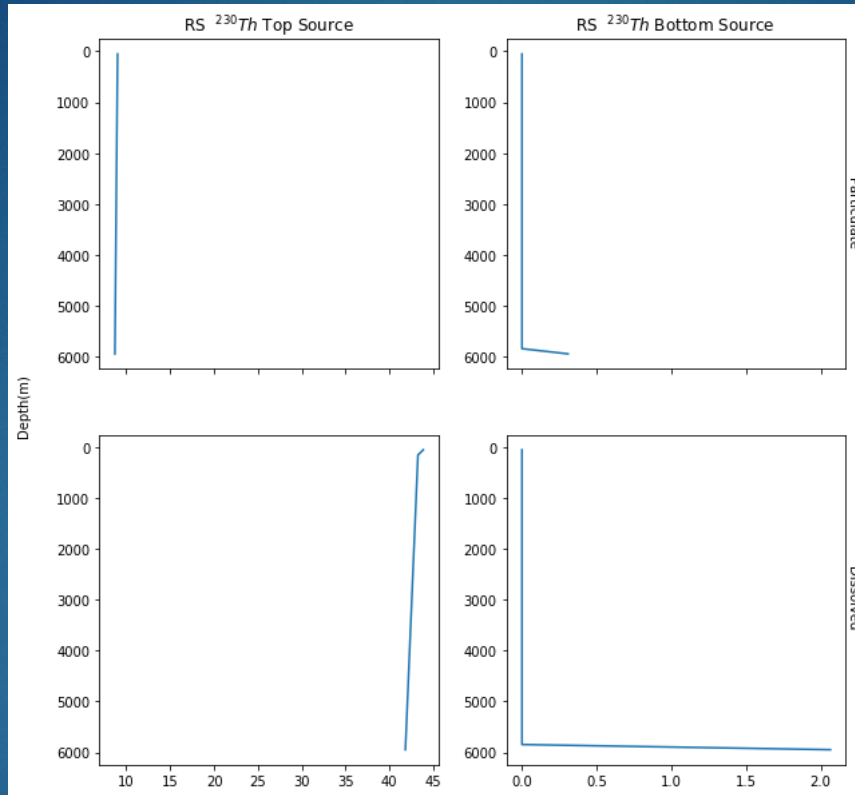
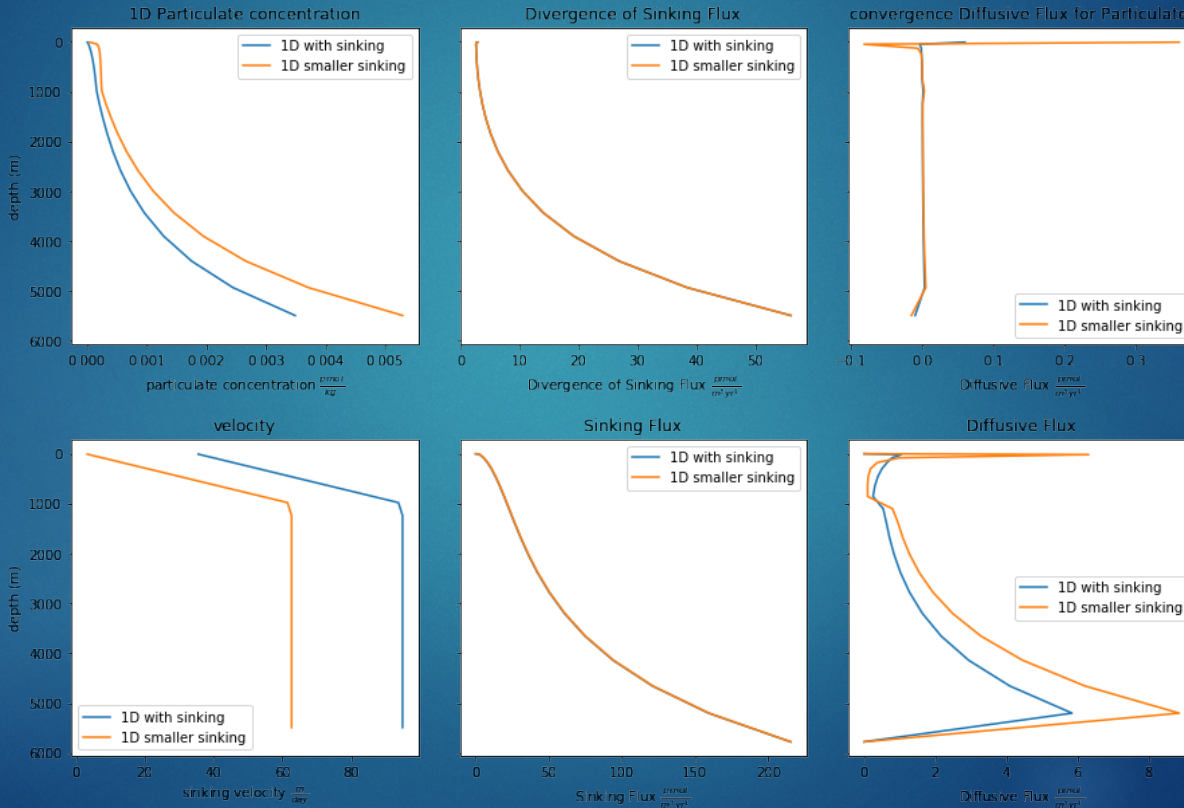


Figure S.3. RS results for a top-only source (left) and a bottom-only source(right). The particle (top) and dissolved (bottom) concentrations.



Fun Math for Fun Math people

Thorium IS equations:

$$\frac{\partial C_d}{\partial t} = Pd - k_p C_d$$

$$C_{d[t+1]} = (Pd - k_p C_{d[t]}) * \Delta t + C_{d[t]}$$

$$\frac{\partial C_p}{\partial t} = k_p C_d - S \frac{\partial C_p}{\partial z}$$

$$C_{p[t+1]} = \left(k_p C_{d[t]} - S \frac{\Delta C_p}{\Delta z} \right) * \Delta t + C_{p[t]}$$

Thorium RS equations:

$$\frac{\partial C_d}{\partial t} = Pd - (\lambda + k_p) C_d + k_{p-1} C_p$$

$$C_{d[t+1]} = (Pd - (\lambda + k_p) C_{d[t]} + k_{p-1} C_{p[t]}) * \Delta t + C_{d[t]}$$

$$\frac{\partial C_p}{\partial t} = (k_p) C_d - (\lambda + k_{p-1}) C_p - S \frac{\partial C_p}{\partial z}$$

$$C_{p[t+1]} = \left((k_p) C_{d[t]} - (\lambda + k_{p-1}) C_{p[t]} - S \frac{\Delta C_p}{\Delta z} \right) * \Delta t + C_{p[t]}$$

Pd	0.303 mol/yr
λ	$9.2 \times 10^{-6} \text{ yr}^{-1}$
S	100 m/yr
k_p	0.52 yr^{-1}
k_{p-1}	2.5 yr^{-1}

Fun Math for Fun Math people

↓ Analytical Solution:

$$C_d = C_{d0} e^{z/\zeta} = C_{dH} e^{-(H-z)/\zeta}$$

where $\zeta = \sqrt{\frac{k_v}{k_p}}$ is the e-folding

depth, H is the depth at the sea floor, $C_{d0} = C_{dH} e^{-H/\zeta}$

↓ Numerical Solution:

$$C_{d[n+1]} = \left(\frac{\Delta F_d}{\Delta z} - k_p C_a + \frac{F_b}{\Delta z} \right) \Delta t + C_{d[n]}$$

$$C_{p[n+1]} = \left(\frac{\Delta F_d}{\Delta z} + k_p * C_d - F_s \right) \Delta t + C_{p[n]}$$

Where $F_d = k_v \frac{\Delta C_p}{\Delta z}$ is the diffusive flux and $F_s = \frac{\Delta(S * C_p)}{\Delta z}$ is the sinking flux

adsorption coefficient as the inverse of residence time $k_p = 1/\tau_R$

$\tau_R = (C_d \rho) / (F_b / H)$, where C_d is the global mean dissolved concentration, $\rho = 1020 \text{ kg/m}^3$ is the average sea water density, F_b is the benthic flux, and $H = 4 \text{ km}$ is the depth of the water column. $C_d = 17.4 \text{ pmol/kg}$

↓ ↓

$$A_r = A_b / A_T$$
$$F_{b_dist} = F_b * A_r$$

History of Neodymium modeling

Bacon and Anderson 1982

Bertram and Elderfield 1993

Henderson et al. 1999

Siddall et al. 2005

Arsouze et al. 2007

Siddall et al. 2008

Arsouze et al. 2009

Rempfer et al. 2011

Gu et al. 2019

Pasquier et al. 2021

Oka et al. 2021

Robinson et al. 2022

Poppelmeier et al. 2022

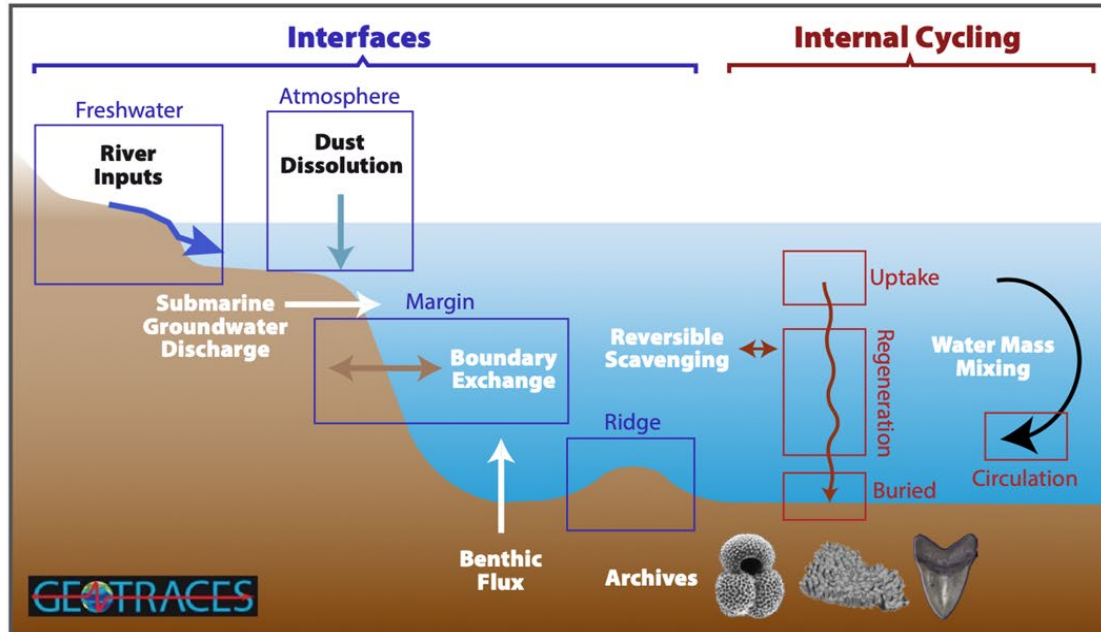
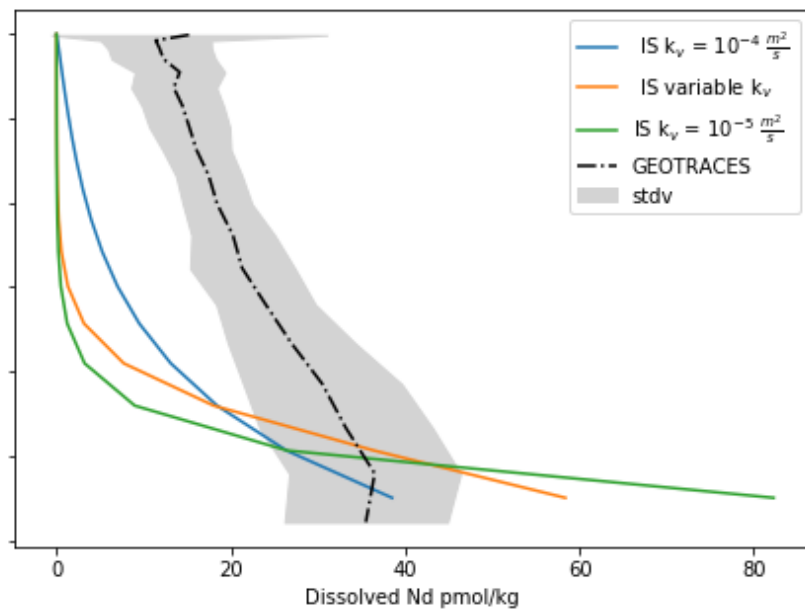
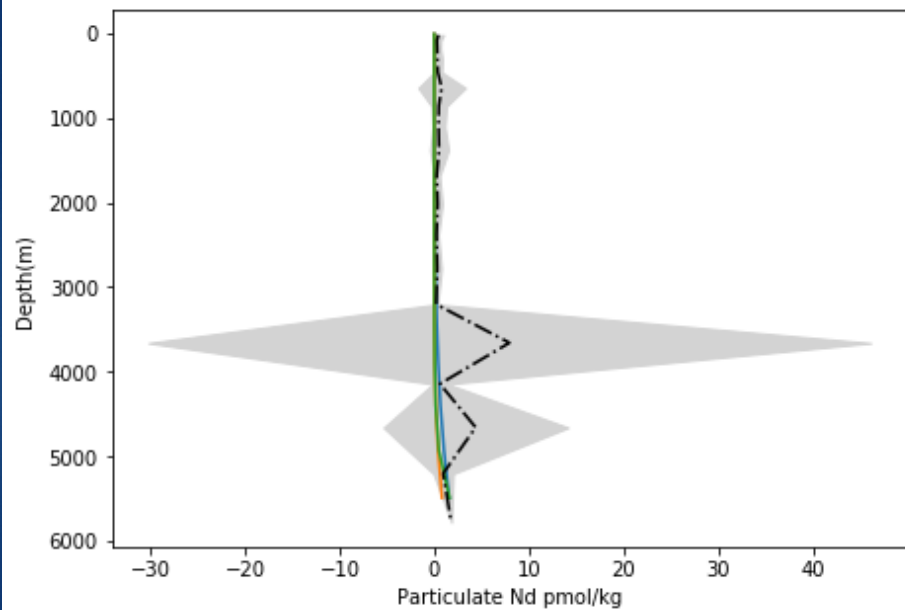


FIGURE 2. Neodymium sources and processes affecting oceanic Nd concentration and isotopic composition as well as possible marine sedimentary records of Nd isotopic composition. Figure based on data from GEOTRACES (<http://www.geotraces.org/>) modified to summarize the oceanic Nd cycle.

k_v Sensitivity



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